



# Use of principal component analysis for the evaluation of the retention behaviour of monoamine oxidase inhibitory drugs on $\beta$ -cyclodextrin column\*

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**Abstract:** The retention of 17 monoamine oxidase inhibitory drugs (propargylamine derivatives) were determined on a  $\beta$ -cyclodextrin polymer ( $\beta$ CDP)-coated silica column using ethanol-0.05 M  $K_2HPO_4$  (6:4 v/v) as the eluent. The relative strength of interaction between the drugs and a water soluble  $\beta$ -cyclodextrin polymer was determined by charge-transfer chromatography carried out on reversed-phase TLC layers. The relationship between capacity factors, physicochemical parameters and inclusion complex forming capacity of the monoamine oxidase inhibitory drugs were evaluated by stepwise regression analysis and by principal component analysis (PCA) followed by two-dimensional nonlinear mapping and varimax rotation. Calculations indicated that the retention of monoamine oxidase inhibitory drugs on  $\beta$ CDP column is mainly governed by their steric and lipophilic parameters. Significant linear correlations were found between the corresponding coordinates of varimax rotation and two-dimensional nonlinear maps proving the suitability of both methods for the reduction of dimensionality of complicated data matrices.

**Keywords:**  $\beta$ -cyclodextrin coated silica; principal component analysis; varimax rotation.

## Introduction

Cyclodextrins (CDs) and various CD derivatives have found growing acceptance and application in many fields of chromatography [1]. They have been used in reversed-phase thin-layer chromatography (RP-TLC) to study their interaction with various bioactive compounds such as barbiturates [2, 3], chlorophenol derivatives [4, 5] etc. CDs modify the effective mobilities of various inorganic ions in isotachopheresis [6], improve separation of peptides in capillary electrophoresis [7] and enhance the efficiency of enantiomeric separation in gas chromatography [8-11]. CDs are used in two different manners in high-performance liquid chromatography either by adding CDs to the eluent [12-15] or by covalently bonding CDs to the silica surface [16-19]. CDs are used either to improve separation of non chiral compounds [20] or to separate enantiomers both in direct and reversed-phase systems [21]. Silica columns with  $\beta$ -cyclodextrin polymer ( $\beta$ CDP) coatings have been prepared recently and their reten-

tion characteristics [22] and their capacity for enantiomer separation have been elucidated [23].

Propargylamine derivatives are selective inhibitors of B-type monoamine oxidase [24, 25], the determination of their lipophilicity [26, 27] and their behaviour in various adsorptive and reversed-phase TLC systems have been recently reported [28].

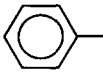
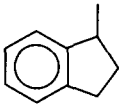
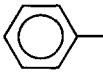
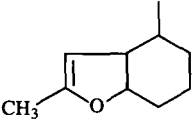
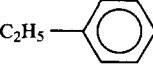
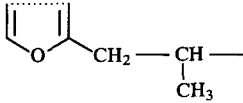
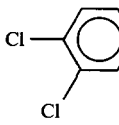
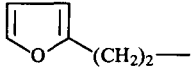
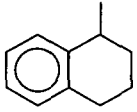
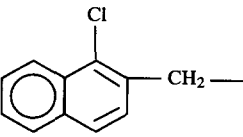
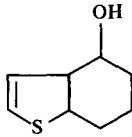
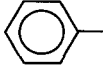
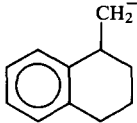
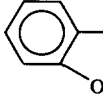
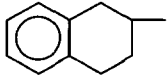
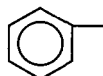
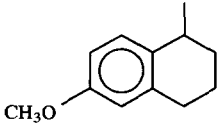
The objectives of our work were to study the retention characteristics of a  $\beta$ -cyclodextrin polymer coated silica column using propargylamine derivatives as solutes, to find relationship between retention behaviour and physicochemical parameters and to compare the applicability of various multivariate mathematical statistical methods for the evaluation of the retention behaviour of these solutes.

## Experimental

The chemical structure of monoamine oxidase inhibitory drugs are compiled in Table 1.

\* Presented at the Fifth International Symposium on Pharmaceutical and Biomedical Analysis, Stockholm, Sweden, September 1994.

**Table 1**  
Chemical structure of monamine oxidase inhibitory drugs

$\begin{array}{c} R_1 - N - CH_2 - C \equiv CH \\   \\ R_2 \end{array}$						
General structure						
No. of compound		R <sub>1</sub>	R <sub>2</sub>	No. of compound	R <sub>1</sub>	R <sub>2</sub>
1	(+)		-CH <sub>3</sub>	10		-CH <sub>3</sub>
2	(-)		-CH <sub>3</sub>			
3			-CH <sub>3</sub>	11		-H
4			-H	12		-CH <sub>3</sub>
5			-CH <sub>3</sub>	13		-CH <sub>3</sub>
6			-CH <sub>3</sub>	14		-CH <sub>3</sub>
7			-CH <sub>3</sub>	15		-CH <sub>3</sub>
8			-CH <sub>3</sub>	16		-CH <sub>3</sub>
9			-C <sub>4</sub> H <sub>7</sub>	17		-CH <sub>3</sub>

*A. Determination of the retention behaviour of drugs by high performance liquid chromatography*

The βCDP coated silica support (patent pending) was prepared at the CYCLOLAB Research and Development Laboratory (Budapest, Hungary). A 25 cm × 4 mm i.d. column was filled in our laboratory with a

Shandon analytical HPLC Packing Pump (Pittsburgh, USA) by the procedure normally used for the filling of reversed-phase columns. The HPLC equipment consisted of a Gilson gradient analytical system GILSON Medical Electronics (Villiers-le-Bell, France) with two piston pumps (Model 302), Detector (Model 116), Rheodyne injector with 20-μl sample

loop (Cotita, California, USA), and a Waters 740 integrator (Milford, Massachusetts, USA). The flow-rate was  $0.8 \text{ ml min}^{-1}$  and the detection wavelength was 240 nm. The eluent was ethanol–0.05 M  $\text{K}_2\text{HPO}_4$  (6:4 v/v). The drugs were dissolved in the eluent at a concentration of  $0.05 \text{ mg ml}^{-1}$ . The retention time of each compound was determined by three consecutive determinations. The capacity factor and the coefficient of variation capacity factor were calculated for each compound.

*B. Determination of the interaction between drugs and a water-soluble  $\beta$ -cyclodextrin polymer by charge transfer thin-layer chromatography*

Silica plates with fluorescence indicator (Silcoflat UV<sub>254</sub>, Kavalier, Brno, Czech Republic) were impregnated with *n*-hexane-paraffin oil (95:5 v/v) by overnight pre-development. Eluents were ethanol–water mixtures, the ethanol concentration varying between 35 and 60 vol% in steps of 5 vol%. To determine the strength of interaction between the drugs and  $\beta$ CD, a water-soluble  $\beta$ -CD polymer (further SCDP) was added to the eluent. Its concentration in the eluent varied between 0 and  $20 \text{ mg ml}^{-1}$ . The SCDP was prepared by cross-linking  $\beta$ -CD monomers with butylene glycol bis(epoxypropyl ether) in aqueous alkaline solution ( $\beta$ CD content 66.04%). The SCDP was purchased from CYCLOLAB Research and Development Laboratory (Budapest, Hungary). We have to emphasize that  $\beta$ CDP and SCDP were prepared with different process of polymerization, therefore, their inclusion forming capacity may also be different. After development the plates were dried at  $105^\circ\text{C}$ , and the spots were detected under UV light and with iodine vapour. Each determination was run in quadruplicate. The  $R_M$  values were calculated by  $R_M = \log(1/R_f - 1)$ . The dependence of  $R_M$  value on the eluent composition was calculated by

$$R_M = R_{M0} + b_1 \cdot C_1 + b_2 \cdot C_2, \quad (1)$$

where  $R_M$  = actual  $R_M$  value of a compound determined at a given ethanol and SCDP concentrations;  $R_{M0}$  =  $R_M$  value of a compound extrapolated to zero ethanol and SCDP concentrations (best estimation of molecular lipophilicity);  $b_1$  = decrease in the  $R_M$  value caused by a 1% increase in the ethanol

concentration in the eluent (related to the specific hydrophobic surface area of drugs);  $b_2$  = decrease in the  $R_M$  value caused by  $1 \text{ mg ml}^{-1}$  change in the concentration of SCDP (indicator of the strength of drug-SCDP complex);  $C_1$  and  $C_2$  = ethanol and SCDP concentrations, respectively.

*C. Calculation of relationships between retention behaviour and physicochemical parameters of propargylamine derivatives*

To elucidate the influence of the physicochemical parameters of drugs on their retention behaviour on  $\beta$ CDP column stepwise regression analysis was applied [29]. The capacity factors of drugs determined on  $\beta$ CDP column were dependent variables and the independent variables were the following physicochemical parameters of drugs:  $\pi$  = Hansch–Fujita's substituent constant characterizing hydrophobicity;  $H - Ac$  and  $H - Do$  = indicator variables for proton acceptor and proton donor properties, respectively;  $M - RE$  = molar refractivity;  $F$  and  $R$  = Swain–Lupton's electronic parameters characterizing the inductive and resonance effect, respectively;  $\sigma$  = Hammett's constant, characterizing the electron-withdrawing power of the substituent;  $Es$  = Taft's constant, characterizing steric effects of the substituent;  $B_1$  and  $B_4$  = Sterimol width parameters determined by distance of substituents at their maximum point perpendicular to attachment; as well as the parameters of equation 1 ( $R_{M0}$ ,  $b_1$  and  $b_2$ ). The number of accepted independent variables was not limited and the acceptance limit was set to 95% significance level.

*D. Comparison of various multivariate – mathematical statistical methods for the study of the retention behaviour of  $\beta$ CDP column Principal component analysis [30]*

The capacity factors of drugs determined on  $\beta$ CDP column, their physicochemical and retention parameters in TLC system (see point C) were the variables and the monoamine oxidase drugs were the observations. The limit of the variance explained was set to 99%. As multidimensional systems cannot be easily evaluated nonlinear mapping technique [31] projects the variables or observations on a two-dimensional plane such a manner that the distances between the points on the plane be approximately the same as their distances in the multidimensional space. The two-dimen-

sional nonlinear map of PC loadings and variables as well as the varimax rotation around two axes were also calculated. The inclusion of both nonlinear mapping technique and varimax rotation in the evaluation was motivated by the consideration that varimax rotation and nonlinear mapping technique are theoretically similar, they calculate and visualize the relative distances between the members of data matrix. To compare their information content linear correlations were calculated between the corresponding coordinates

$$Y_{1-2} = a + b.X_{1-2}, \quad (2)$$

where  $Y_{1-2}$  = coordinates of nonlinear map;  
 $X_{1-2}$  = coordinates of varimax rotation.

## Results and Discussion

### A. High-performance liquid chromatography

The retention order of propargylamine derivatives does not follow the lipophilicity order (Fig. 1). This finding suggests that the retention characteristics of  $\beta$ CDP column deviate from those of alkyl bonded silica columns where the retention order is determined by the lipophilicity order of the solutes. We assume that the following interactions may influence the retention behaviour of solutes on  $\beta$ CDP column.

1. *Interactions of solutes with CD cavity.* These interactions are determined by the size of the guest molecules and their lipophilicity.

The steric parameters define the capacity of the guest molecule to enter in the CD cavity and the lipophilicity of the guest molecule determines the strength of interactions with the hydrophobic inner surface of the CD cavity.

2. *Polar interactions between the solutes and surface of  $\beta$ CDP support.* Hydrophilic forces can bind the polar substructure of solutes to the hydroxyl groups on the  $\beta$ CDP surface or to the free silanol groups of silica support not covered by  $\beta$ CDP.

The retention of solutes is probably determined by the interplay of the various forces discussed above.

The  $\log k'$  values and the coefficients of the variation are listed in Table 2. The  $\log k'$  values show high diversity indicating that the drugs can be successfully separated on this column. The coefficients of variation is low indicating the good reproducibility of retention time on  $\beta$ CDP column.

### B. Charge-transfer thin-layer chromatography

The parameters of equation 1 are compiled in Table 3. Compounds 1 and 2 were omitted from the calculations because they exhibited elongated spots in the eluents resulting in the inaccurate determination of their retention. Equation 1 fits well to the experimental data, the significance level being over 95% (see calculated  $F$  values) in each instance. The ratio of variance explained by the independent variables varied between 51 and 96% (see  $r^2$  values). Both lipophilicity value ( $R_{M0}$ ) and the

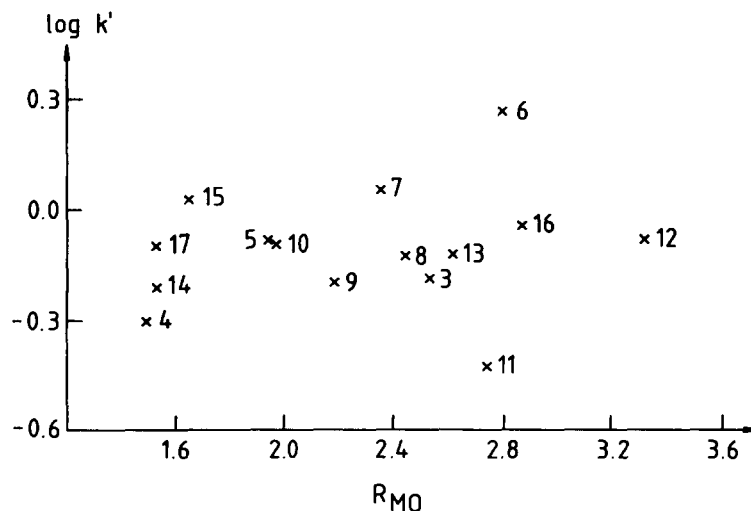


Figure 1 Relationship between the retention of monoamine oxidase inhibitory drugs on the  $\beta$ -cyclodextrin polymer-coated column and the measured hydrophobicity value ( $R_{M0}$ ). Numbers refer to drugs in Table 1.

**Table 2**

Retention of monoamine oxidase inhibitory drugs on  $\beta$ -cyclodextrin polymer-coated silica column. Eluent: ethanol–50 mM  $K_2HPO_4$  (6:4, v/v). Numbers refer to monoamine oxidase inhibitory drugs in Table 1

No. of compound	log $k'$	
	mean	coefficient of variation %
1	-0.127	0.24
2	-0.127	0.31
3	-0.182	0.52
4	-0.303	0.22
5	-0.082	0.36
6	0.270	0.29
7	0.053	0.23
8	-0.124	0.37
9	-0.191	0.21
10	-0.089	0.47
11	-0.423	0.55
12	-0.077	0.61
13	-0.122	0.43
14	-0.209	0.79
15	0.029	0.89
16	-0.038	0.95
17	-0.095	0.54

specific hydrophobic surface area ( $b_1$ ) of the drugs differ considerably, indicating that these parameters can be separately included in the future Quantitative Structure Activity Relationship calculations.

#### C. Relationship between the retention behaviour and physicochemical parameters of drugs

Stepwise regression analysis indicated that the retention of monoamine oxidase inhibitory drugs significantly depend on their calculated lipophilicity

$$\log k' = -0.60 + (0.13 \pm 0.02) \cdot \pi \quad (3)$$

$$r_{\text{calc.}} = 0.8639 \quad r_{99.9\%} = 0.7246$$

$$\text{Standard error of the estimate} = 0.09$$

Equation 3 clearly shows that the calculated hydrophobicity of solutes has the highest impact on their retention that is the retention mechanism of  $\beta$ CDP support may be similar to that of the traditional alkyl bonded silica phases. However, the relatively low ratio of variance explained (74%) indicates that factors other than molecular hydrophobicity may also have a significant influence on the retention. The results of stepwise regression analysis slightly contradict the results in Fig. 1. This discrepancy can be tentatively explained by the supposition that slight difference between the measured and calculated hydrophobicity values.

#### D. Comparison of various multivariate methods for the evaluation of the retention data

The results of PCA are summarized in Table 4. Five principal components explain the majority of variance indicating that the 15 original variables can be substituted by five background (composite) variables without considerable loss of information. Unfortunately, PCA does not prove the existence of such composite variables as concrete physicochemical entities only indicates their mathematical possibility. The log  $k'$  values — together with the measured hydrophobicity, complex forming capacity, steric and electronic parameters of drugs — have high loading in the second PC indicating the marked influence of these parameters on the mode of retention of the  $\beta$ CDP support. The distribution of variables on the two-dimensional nonlinear map of PC loadings supports our previous conclusions (Fig. 2), the log  $k'$  values form a loose cluster with the measured hydrophobicity, complex forming capacity, steric and electronic parameters of drugs. This finding indicates again the mixed retention mechanism of the  $\beta$ CDP support.

The propargylamine derivatives do not form distinct clusters on the two-dimensional non-linear map of principal component variables (Fig. 3). This result suggests that the character and position of each substituent may have similar impact on their retention behaviour.

Good linear correlation were found between the corresponding coordinates of nonlinear map and varimax rotation ( $r = 0.9963$  and  $r = 0.9295$  for the first and second coordinates, respectively) suggesting that the information content of the methods is similar. *We have to stress that the conclusion discussed above is not the result of theoretical considerations and hence is valid only for this data set.* We assume that the generalization of the conclusion may lead to severe misinterpretation.

It can be concluded from the data that  $\beta$ CDP support shows retention behaviour different from that of alkyl bonded silicas, the steric and electronic parameters of solutes markedly influencing the retention. Principal component analysis followed either with nonlinear mapping technique or with varimax rotation proved to be a useful tool for the elucidation of the effect of various molecular parameters on the retention.

**Table 3**

Relationship between the lipophilicity ( $R_{M0}$ ) of monoamine oxidase inhibitory drugs and the concentration of ethanol ( $C_1$ ) and water-soluble  $\beta$ -cyclodextrin polymer ( $C_2$ ) in the eluent. Numbers refer to monoamine oxidase inhibitory drugs in Table 1. Compound 1 and 2 were omitted from calculations ( $n = 18$ )

$$R_M = R_{M0} + b_1 \cdot C_1 + b_2 \cdot C_2$$

Parameter	No. of compound						
	3	4	5	6	7	8	9
$R_{M0}$	2.53	1.50	1.94	2.80	2.35	2.44	2.17
$-b_1 \cdot 10^{-2}$	4.82	3.66	4.26	5.24	3.00	4.43	1.62
$S_{b1} \cdot 10^{-3}$	2.7	3.0	2.8	3.5	7.3	3.4	4.4
$b_1$ (%)	80.13	85.51	80.00	81.25	69.77	—	—
$-b_2 \cdot 10^{-2}$	1.22	6.15	1.00	1.42	1.39	—	—
$S_{b2} \cdot 10^{-3}$	2.8	2.7	2.9	3.6	3.7	—	—
$b_2$ (%)	19.87	14.49	20.00	18.75	30.23	—	—
$r^2$	0.9596	0.9179	0.9455	0.9415	0.8294	0.9099	0.9523
$F_{calc}$	178.43	78.01	130.22	120.83	14.59	161.68	14.70

**Table 4**

Similarities and dissimilarities between the physicochemical parameters of monoamine oxidase inhibitory drugs and their retention on  $\beta$ -cyclodextrin polymer coated silica column. Results of principal component analysis

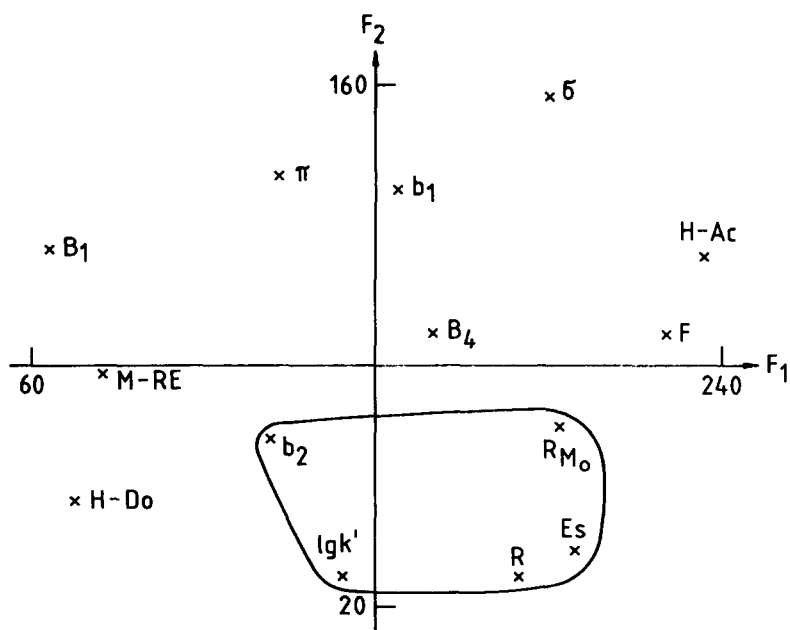
No. of component		Eigenvalue	Total explained variance (%)				
1		5.45	36.38				
2		3.11	57.15				
3		2.64	74.53				
4		1.47	84.56				
5		0.93	90.79				

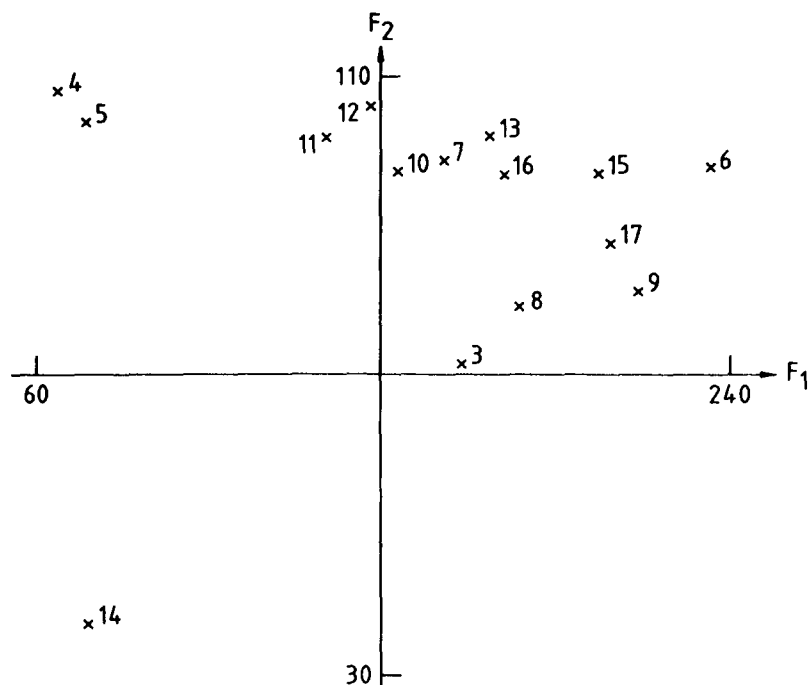
		Principal component loadings				
		1	2	3	4	5
1	$\pi$	0.12	0.14	-0.55	0.18	0.70
2	<i>H-Ac</i>	0.89	-0.38	0.17	0.05	0.01
3	<i>H-Do</i>	-0.82	0.29	0.29	-0.25	-0.02
4	<i>M-Re</i>	-0.61	0.34	0.17	0.50	0.20
5	<i>F</i>	0.81	-0.15	0.43	0.27	-0.01
6	<i>R</i>	0.40	0.68	-0.18	-0.52	0.02
7	$\sigma$	0.53	-0.46	-0.60	0.15	-0.21
8	<i>Es</i>	0.45	0.62	-0.30	-0.47	0.16
9	$B_1$	-0.82	-0.07	0.08	0.17	0.29
10	$B_4$	0.42	0.36	0.77	0.19	0.17
11	$b_1$	0.22	-0.06	0.90	-0.20	0.06
12	$b_2$	0.19	-0.59	-0.28	0.61	-0.24
13	$R_M$	0.79	-0.51	0.04	0.17	-0.04
14	$\log k'$	0.11	0.90	0.03	0.17	-0.15

**Table 3**  
Continued

No. of compound							
10	11	12	13	14	15	16	17
1.97	2.74	3.33	2.61	1.53	1.65	2.85	1.53
3.85	5.32	5.42	3.99	3.81	4.82	4.42	4.17
4.2	4.41	3.7	3.4	4.3	3.3	3.3	3.0
77.00	79.00	78.26	78.00	81.02	81.01	77.19	73.21
1.10	2.10	1.57	1.15	8.94	1.65	1.37	1.59
4.4	4.6	3.88	3.6	3.9	3.5	3.53	3.1
23.00	21.00	21.74	22.00	18.98	18.99	22.81	26.79
0.8934	0.9091	0.9403	0.9045	0.9525	0.9360	0.9253	0.9301
43.28	75.08	118.18	71.07	40.47	109.71	92.92	99.83



**Figure 2**  
Two-dimensional nonlinear map of PC loadings. Number of iteration: 86. Maximum error:  $6.29 \times 10^{-3}$ . For symbols see Experimental.



**Figure 3**

Two-dimensional nonlinear map of PC variables. Number of iteration: 250. Maximum error:  $4.81 \times 10^{-3}$ . Numbers refer to drugs in Table 1.

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